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Structure of 1,3,5-Trimethyl-2,2,4,4,6,6-hexakis(methylamino)cyclotrisilazane

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Abstract. C₉H₃₃N₉Si₃, $M_r = 351.5$, monoclinic, $P2_1/n$, a = 8.737 (2), b = 14.336 (3), c = 16.120 (3) Å, $\beta =$ $V = 2003 (4) \text{ Å}^3,$ $D_r =$ 97·20 (6)°. Z = 4, 1.1661 g cm^{-3} . λ (Mo K α) = 0.71069 Å, $\mu =$ 2.39 cm^{-1} , F(000) = 768, room temperature, final R = 0.0573 for 1935 observed reflections and 191 variables, $F(hkl) > 4\sigma(F)$. The six-membered Si₃N₃ ring is planar with a mean value of 1.703 Å for the Si-N bond lengths. The methyl groups attached to the ring N atoms are nearly coplanar with the ring and molecular symmetry (excluding H atoms) nearly fulfills $\overline{6}2m$. The molecules are stacked in the manner of coin rolls along [101].

Experimental. C₉H₃₃N₉Si₃ was synthesized by reaction of SiCl₄ and CH₃NH₂ in *n*-pentane. Transparent crystals were grown from the vapor phase. Crystal $0.4 \times 0.5 \times 0.5$ mm. Automated Enraf-Nonius CAD-4 diffractometer, graphite-monochromated Mo K α radiation. 25 centered reflections within 4 < $\theta < 12^{\circ}$ used for determining lattice parameters. Data corrected for Lorentz and polarization effects, absorption ignored. $2\theta_{\text{max}} = 50^{\circ}$, range of *hkl*: $0 \le h$

$\leq 10, 0 \leq k \leq 17, -19 \leq l \leq 19$. Two check reflec-
tions measured every 200 reflections showed no sig-
nificant intensity variation over data collection.
$\omega/2\theta$ -scan technique. Total of 3940 reflections meas-
ured, 2857 unique (non-zero), 1935 observed with
$F(hkl) > 4\sigma(F)$. Structure solved by direct methods,
full-matrix least-squares refinement on F of 191
parameters (SHELX76, Sheldrick, 1976) on a
VAX11/750 computer. Anisotropic thermal param-
eters for non-H atoms, H atoms allowed to ride at
fixed distance on C and N atoms, refined isotropi-
cally. $R = 0.057$, unit weights, $R_{int} = 0.012$, $(\Delta/\sigma)_{max}$
= 0.001, $\Delta \rho_{\text{max}} = 0.24$, $\Delta \rho_{\text{min}} = -0.26$ e Å ⁻³ , atomic
scattering factors those incorporated in SHELX76.

Table 1 lists atomic positional and equivalent isotropic thermal parameters, Table 2 interatomic distances and valence angles.* Fig. 1 shows a

Table 2. Bond lengths (Å) and angles (°) Si(1)—N(1) 1.707 (5) Si(2)—N(7) 1.715 (5) N(3)—C(3) 1.483 (7)

Table	1.	Fractional	coordinates	and	equivalent	iso-
tro	pic	temperature	coefficients	for r	ion-H atom	S

$U_{eq} = (1/3) \sum_{i} \sum_{j} U_{ij} a_i^* a_j^* \mathbf{a}_i \cdot \mathbf{a}_j.$				Si(1) - N(2) Si(1) - N(4)	1·695 (5)	Si(3) - N(1) Si(3) - N(3)	1·702 (5) 1·696 (5)	N(4) - C(4) N(5) - C(5)	1.428 (9)	
	x	v	z	$U_{\rm res}$ (Å ²)	Si(1) - N(5)	1.715 (5)	Si(3)—N(8)	1.710 (4)	N(6)-C(6)	1.461 (8)
Si(1)	1.1656 (1)	0.8154 (1)	0.6642(1)	0.0476 (5)	Si(2)—N(2)	1.711 (5)	Si(3)—N(9)	1·701 (6)	N(7)-C(7)	1.438 (9)
Si(2)	0.9749(1)	0.6372 (1)	0.6931 (1)	0.0482 (6)	Si(2)—N(3)	1.709 (5)	N(1) - C(1)	1·484 (8)	N(8)—C(8)	1·444 (9)
Si(3)	0.8454 (1)	0.8338 (1)	0.7297 (1)	0.0506 (5)	Si(2)—N(6)	1.709 (5)	N(2)—C(2)	1.501 (8)	N(9)—C(9)	1.443 (9)
N(1)	1.0174 (5)	0.8713 (3)	0·7030 (3)	0.0594 (19)						
N(2)	1.1212 (4)	0.7004 (3)	0.6592 (3)	0.0491 (17)	N(2)—Si(1)—	N(1) 1	107-1 (2)	N(9)—Si(3)	—N(8)	101-3 (3)
N(3)	0.8447 (5)	0.7157 (3)	0.7224 (3)	0.0596 (19)	N(4)—Si(1)—	N(1) 1	13.7 (3)	C(1) - N(1)	— Si(1) .	113-1 (4)
N(4)	1.1994 (6)	0.8551 (4)	0.5688 (3)	0.0785 (24)	N(4)—Si(1)—	N(2) 1	10.3 (3)	C(1) - N(1)	—Si(3)	113.7 (4)
N(5)	1.3353 (5)	0.8412 (3)	0.7255 (3)	0.0617 (20)	N(5)—Si(1)—	N(1) 1	l09·1 (3)	C(2)-N(2)	—Si(1)	113.8 (4)
N(6)	1.0299 (5)	0.5627 (3)	0.7742 (3)	0.0618 (22)	N(5)—Si(1)—	N(2) I	14·4 (3)	C(2)—N(2)-	—Si(2)	113-3 (4)
N(7)	0.8979 (6)	0.5621 (3)	0.6162 (3)	0.0725 (25)	N(5)—Si(1)—	N(4) 1	l02·3 (3)	C(3)—N(3)	—Si(2)	113-4 (4)
N(8)	0.6883 (5)	0.8796 (3)	0.6695 (3)	0.0657 (20)	N(3)—Si(2)—	N(2) 1	106-8 (2)	C(3)—N(3)-	—Si(3)	113-9 (4)
N(9)	0.8183 (6)	0.8735 (4)	0.8262 (3)	0.0847 (27)	N(6)—Si(2)—	N(2) 1	15-2 (2)	C(4)—N(4)	—Si(1)	124.9 (5)
C(1)	1.0344 (8)	0.9744 (6)	0.7043 (6)	0.1132 (45)	N(6)—Si(2)—	N(3) 1	l09·5 (3)	C(5)—N(5)	—Si(1)	122-1 (4)
C(2)	1 2426 (7)	0.6407 (4)	0.6275 (5)	0.0855 (32)	N(7)-Si(2)-	N(2) 1	10.0 (2)	C(6)—N(6)	—Si(2)	122.6 (5)
C(3)	0.7035 (7)	0.6715 (5)	0.7467 (6)	0.1142 (42)	N(7)—Si(2)—	N(3) 1	13.5 (3)	C(7)—N(7)	—Si(2)	122-9 (5)
C(4)	1.0988 (10)	0.8415 (7)	0.4929 (5)	0.1259 (49)	N(3)—Si(3)—	N(1) 1	107-1 (2)	C(8)—N(8)	—Si(3)	121-4 (5)
C(5)	1 3574 (7)	0.8231 (6)	0.8144 (4)	0.0902 (35)	N(7)—Si(2)—	N(6) 1	l02·0 (3)	C(9)—N(9)	—Si(3)	126-0 (5)
C(6)	1.0784 (9)	0.5953 (5)	0.8593 (4)	0.0904 (34)	N(8)—Si(3)—	N(1) 1	14-0 (3)	Si(3)—N(1)	—Si(1)	132.7 (3)
C(7)	0.8455 (10)	0.5920 (6)	0.5323 (5)	0.1106 (41)	N(8)—Si(3)—	N(3) 1	110·4 (3)	Si(2)—N(2)	—Si(1)	132-5 (4)
C(8)	0.6626 (8)	0.8644 (7)	0.5803 (4)	0.1127 (44)	N(9)—Si(3)—	N(1) 1	10.7 (3)	Si(3)—N(3)	—Si(2)	132-8 (3)
C(9)	0.9274 (10)	0.8666 (7)	0.9009 (5)	0·1267 (44)	N(9)—Si(3)—	•N(3) 1	l 13·4 (3)			

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^{*} Lists of structure factors, anisotropic thermal parameters and H-atom parameters have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 52644 (17 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.



Fig. 1. The molecular structure of the title compound.

perspective view of the molecule, Fig. 2 crystal packing.

Related literature. With respect to the Si_3N_3 core, related silazanes are 2,2,4,4,6,6-hexaisopropylcyclotrisilazane (Klingebiel & Vater, 1983), 2,4,6-tri-*tert*butyl-2,4,6-trifluorocyclotrisilazane (Clegg, Sheldrick & Stalke, 1984b) and 2,2,4,4,6,6-hexa-*tert*-butylcyclotrisilazane (Clegg, Sheldrick & Stalke, 1984a), all showing a planar six-membered ring in agreement with our results.

The reactions of $SiCl_4$ with CH_3NH_2 in the gas phase (Drake & Westwood, 1971), without solvent at room temperature (Hagen & Callaway, 1972) and in petroleum ether (Adrianov, Il'in, Talanov, Isakova &



Fig. 2. Stereoplot of the unit cell. H atoms are omitted.

Sidorenko, 1976) have been reported in the literature. Based on an elementary analysis, Adrianov *et* al. (1976) suggested for one of their products the same composition as that of the title compound; however, no information on the molecular structure was given.

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9-(2,4-Cyclopentadienylidene)bicyclo[3.3.1]nonane, a Ring-Strained Pentafulvene

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Abstract. 9-(2,4-Cyclopentadien-1-ylidene)bicyclo-[3.3.1]nonane, C₁₄H₁₈, $M_r = 186\cdot3$, monoclinic, $P2_1/c$, $a = 10\cdot6088$ (11), $b = 12\cdot251$ (3), c =9·2343 (14) Å, $\beta = 111\cdot23$ (1)°, $V = 1118\cdot7$ (6) Å³, Z =4, $D_x = 1\cdot106$ Mg m⁻³, λ (Cu K α) = 1·54184 Å, $\mu =$ $0\cdot425$ mm⁻¹, F(000) = 408, T = 295 K, $R = 0\cdot042$ for 1636 observations with $I > 3\sigma(I)$ (of 2300 unique data). The bicyclo[3.3.1]nonane adopts a twin-chair conformation. The bond angle of the C=C exocyclic to the cyclopentadienylidene ring is 110.95 (9)°. This bond angle is very close to the corresponding bond angle in 2-(2,4-cyclopentadien-1-ylidene)adamantane or adamantylidenefulvene. The cyclopentadienylidene ring is planar, with maximum deviation of 0.005 (2) Å.

Experimental. The title compound was prepared by condensing bicyclo[3.3.1]nonan-9-one and 1,3-© 1990 International Union of Crystallography

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